**Chemistry 201: General Chemistry for Engineers, Winter 2019**

| **Instructors** | **Office** | **Email** | **Office Hours** |
| --- | --- | --- | --- |
| Marci Moling | MH-210 | molingm@linnbenton.edu | MTW 12:30-2:00 pmR 2:00-4:00 pm |

**Lecture:** MW 11:00 am-12:20 pm and F 11:00-11:50 am in MH-208

# **Laboratory:**

Omid Sadeghi; sadegho@linnbenton.edu R 8 am, 11 am, or 2 pm, MH-206

# **Outcomes:**

* Differentiate the historical developments leading to the development of the atomic theory and the Periodic Table.
* Solve scientific problems with quantitative methods using dimensional analysis and/or algebra regarding unit conversions, stoichiometry, gas laws, and thermochemistry.
* Apply chemical principles associated with chemical and physical changes and properties of matter, nomenclature, chemical reactions, thermochemistry, the kinetic theory of a gas, and quantum theory.
* Work safely in a laboratory environment while observing and accurately recording measurements related to chemical phenomena.

**Minimum Requirements:** MTH 095 and any one of the following: a passing score on the chemistry entrance exam, or CH 150 with a grade of “C” or better, or CH 121 with a grade of “C” or better, or CH 112 with a grade of “C” or better. **Corequisite:** MTH 111.

# **Required Materials:**

*Chemistry: The Molecular Nature of Matter and Change,* 8th Ed., Silberberg

*Chemistry 201 Lab Manual*

Carbonless Lab Notebook

Non-graphing/non-programmable Scientific Calculator

**Optional Materials:** 1. Lab coat 2. Personal Safety Goggles

**Science Help Desk:** The Science Help Desk is located on the first floor of Madrone Hall in the atrium area. The Help Desk is manned approximately 20 hours per week. Hours of the Help Desk are posted in the Help Desk area.

**Calculator Policy:**  Students will be required to use a non-graphing/non-programmable scientific calculator for quizzes and/or exams.  Department approved calculators are:  TI 30xa, TI 30X IIs, Casio fx-260, or HP 10s.  If a student does not wish to purchase one of these calculators the department will provide either a Casio fx-260 or HP 10s for use on exams and/or quizzes.

**Attendance and Classroom Decorum:** Class attendance is very important to the learning of chemistry. Students are expected to attend class regularly and on time. Entering the classroom late or leaving before the class ends is distracting to students and your instructor. **There is NO cell phone use in the classroom allowed.** The use of a laptop computer during lecture class is approved for CH 201 lecture material **only**.

**Homework Problem Sets**:To succeed in chemistry, like learning a foreign language, you should study and practice every day. As material is covered you will find the problems are easier to work and not as time consuming as if they are attempted just before the due date. Keep in mind a typical science course takes **3-4 hrs of work per week outside of class for every credit hour**.

**Quizzes:** You will be given a take home quiz for each chapter. It will be due the at the beginning of the next lecture. Quizzes will reflect material from the assigned homework. The quiz problems are good practice for exams and assist with keeping students up-to-date with material. No make-up quizzes are given. The lowest quiz score will be dropped.

**Exams:** All exams are given in class. Students who have conflicts with exam days due to other College functions, illness, or family emergencies must contact the instructor prior to the exam. Documentation of the College function, illness and/or family emergency must be provided to schedule a make-up exam.

**Laboratory Reports**: Lab reports are due at the beginning of YOUR next lab session after the completion of the experiment. Late lab reports receive a 10% per day mark down. One lowest lab report score is dropped. You must receive at least 70% of the total lab points to pass the course regardless of passing the lecture. **No make-up labs will be given.** **Also, if you miss more than three labs or turn in fewer than five reports you will not receive a passing grade for the course. This is a lab class and in order to pass the course you must pass the laboratory component.**

**Prelab Questions: Be sure to check the syllabus for which lab is assigned for a particular week.** Most lab experiments described in the manual have prelab questions. Many of these questions are designed to emulate the laboratory experiment that is about to be performed. By answering these questions BEFORE the lab period students are able to understand and perform the experiment more effectively. Prelab questions should be done on separate sheets of paper and are due **within the first 5 minutes** of the lab period. The prelab assignments are worth from one to five points of the lab report grade. No late prelabs are accepted.

**Extra Credit:** If you complete all the quizzes your lowest quiz score will be used as extra credit.

# **Grading:**

| 3 Exams (3 @ 100 pts each) | 300 pts |
| --- | --- |
| Final Exam | 130 pts |
| Quizzes (6 @ 20 pts each) | 120 pts |
| Lab |  |
| * 7 Lab Reports (7 @ 20 pts each)
 | 140 pts |
| * Nomenclature Worksheet
 | 20 pts |
| * Lab Final Exam
 | 40 pts |
| Total  | 750 pts |

# **Course Grade:**

|  |  |  |  |  |
| --- | --- | --- | --- | --- |
| 90-100% A | 80-89% B | 70-79% C | 60-69% D | 0-59% F |

An incomplete grade (I) may be given at the discretion of the instructor. However, a student must have a passing grade at the time an incomplete is assigned.

# **Drop/Withdraw Policy:**

If you are withdrawing from the class you must file a Schedule Change Form with Registration or use WebRunner. If you formally drop the class **by Monday of the second week of the term**, you will receive a tuition refund. If you withdraw after the Monday of the second week of instruction through the seventh week a **‘W’** will show up on your transcript. No withdrawals are allowed after the end of the seventh week. An instructor may not assign a “W” grade.

If you received financial aid or veteran’s benefits PLEASE talk with associates at the appropriate office to determine what effects on eligibility dropping a course will have. Don’t jeopardize your eligibility!! You can contact the Financial Aid Office by calling (541) 917-4850 or by visiting the Financial Aid Office in Takena Hall.

If you stop attending the course without formally withdrawing you will continue to accumulate grades (zeroes for all assignments not turned in) and will receive the grade assigned by the instructor. You will also be held accountable for all charges on your account.

# **Academic Integrity:**

“An instructor has the right to issue a grade of F for the course in which the instructor has reason to believe the student has cheated. A student has the right to appeal such action in accordance with the Students’ Rights, Responsibilities and Conduct Policy.” The preceding statement is Administrative Rule No. 7030-01.

# **Center for Accessibility Resources:**

You should meet with your instructor during the first week of class if:

1. You have a documented disability and need accommodations.

2. Your instructor needs to know medical information about you.

3. You need special arrangements in the event of an emergency.

If you have documented your disability, remember that you must make your request for accommodations through the Center for Accessibility Resources Online Services web page every term in order to receive accommodations. If you believe you may need accommodations but are not yet registered with CFAR, please visit the CFAR website at www.linnbenton.edu/cfar for steps on how to apply for services or call 541-917- 4789.

# **LBCC Comprehensive Statement of Nondiscrimination:**

LBCC prohibits unlawful discrimination based on race, color, religion, ethnicity, use of native language, national origin, sex, sexual orientation, gender, gender identity, marital status, disability, veteran status, age, or any other status protected under applicable federal, state, or local laws. For further information see Board Policy P1015 in our Board Policies and Administrative Rules.

#  **Lecture and Lab Schedule:**

**\*\*Note: This schedule of topics, homework due dates, and exam dates are subject to change.**

| **Week No.** | **Mon.** | **Wed.** | **Fri.** | **Laboratory** |
| --- | --- | --- | --- | --- |
| **Week 1**1/7-1/11 | 1.1-1.3 | 1.4, 2.1**Quiz 1** | 2.1-2.2 | Safety, Lab Format,Sig Fig Review, Section 2.8, Nomenclature Worksheet (Due next lab period) |
| **Week 2**1/14-1/18 | 2.3-2.5 | 2.5-2.7**Quiz 2** | 3.1-3.2 | Uncertainties in MeasurementBalance InstructionsExpt 1: Are All Crayons Created Equal?  |
| **Week 3**1/21-1/25 | **Holiday****No Class** | **Exam 1** **(Ch 1 & 2)** | 3.2-3.3  | Expt 2: The Formula of a Hydrate |
| **Week 4**1/28-2/1 | 3.3-3.4  | 3.4**Quiz 3** | 4.1 | Expt 3: Identification of Household Chemicals |
| **Week 5**2/4-2/8 | 4.2-4.3 | 4.3-4.4 | 4.5**Quiz 4** | Expt 4: Generic vs. Name Brand Vinegar |
| **Week 6**2/11-2/15 | 4.5, 5.1-5.2 | **Exam 2** **(Ch 3 & 4)** | 5.2**Quiz 5** | Expt 5: Chemical Reactions  |
| **Week 7**2/18-2/22 | **Holiday****No Class** | 5.3-5.5 | 5.5, 5.6 | Expt 6: The Universal Gas Constant |
| **Week 8**2/25-3/1 | 5.6, 6.1-6.2  | 6.3-6.4  | 6.5**Quiz 6** | Expt 7: The Specific Heat of Pennies  |
| **Week 9**3/4-3/8 | 6.6, 7.1 | **Exam 3** **(Ch 5 & 6)** | 7.1-7.2 | Expt 8: Hess’s Law |
| **Week 10**3/11-3/15 | 7.2-7.3 | 7.3-7.4**Quiz 7** | 7.4 | Lab Final Exam |
| **Week 11**3/18-3/20 |  | **Final Exam****10-11:50 am** |  |  |

# **Homework Problems:**

Ch 1: 30, 36, 39, 47, 48, 54, 56, 66, 76, 79, 81, 85

Ch 2: 21, 28, 40, 48, 50, 85, 89, 99, 105, 107, 110, 132, 138, 141

Ch 3: 13, 17, 21, 23, 31, 37, 43, 45, 57, 59, 61, 62, 69, 73, 75, 84, 88, 91, 132, 138

Ch 4: 14, 15, 19, 23, 25, 27, 31, 42, 45, 46, 48, 54, 73, 76, 84, 87, 91, 94, 95, 97

Ch 5: 7, 9, 14, 21, 25, 27, 29, 32, 35, 37, 39, 46, 48, 52, 53, 60, 63, 68, 76, 80, 91

Ch 6: 9, 10, 11, 13, 23, 38, 40, 41, 49, 55, 60, 68, 80, 85, 96, 102

Ch 7: 8, 10, 15, 24, 30, 32, 42, 50, 52, 56, 60, 64, 73, 81, 92